The Ideal Bait Tank Environment: How to keep your bait healthy.

Shad become **extremely** stressed when they are caught and handled. Stress causes a fish's metabolism as well as adrenaline, cortisol and blood sugar levels to skyrocket. Some physiological responses to stress include:

- Loss of slime and scales
- Increased oxygen demand
- Increased ammonia production
- Increased carbon dioxide production
- Loss of salts

Our goal is to minimize these adverse reactions to stress by maintaining optimum water quality conditions in our bait tanks.

The Basics

- More water volume is always better.
- Use cool water BUT not more than 5-7 degrees cooler than the water in which the bait was caught. Cool water holds more oxygen than warm water.
- 1 medium shad per gallon of water in 60* water,
 ½ shad per gallon in 80*water. Fewer lively baits are better than many weak baits.
- Be gentle to your bait. Don't empty your cast net on the floor of the boat.

Oxygen

- Commercially available bait tanks equipped with aerators do a satisfactory job oxygenating the water as long as the tank is not overloaded or too hot.
- Supplemental oxygen provided with a tank, regulator and diffuser set-up is beneficial in hot weather or with a heavily loaded tank.
- Check your tank frequently to make sure the pump is working and the aerator is not clogged.

Ammonia

Ammonia is a highly toxic waste product excreted by fish. The rate of ammonia excretion is very high for an hour after catching bait.

- Ammonia CAN NOT be removed by aeration.
- Small, frequent water changes, especially in the first hour after catching bait reduce ammonia build up.
 Large, infrequent water changes will shock the bait.
- Chemicals which bind to ammonia and render it harmless are available. Use them at 10x the recommended dosage.

Carbon Dioxide (CO₂)

Fish "exhale" CO₂ just like humans. CO₂ acidifies the bait tank water and fish blood and decreases blood's oxygen carrying capacity. Fortunately, CO₂ degasses when water is exposed to air.

- Aerate and agitate! There is no such thing as too many bubbles. There is such a thing as too much current.
- Leave the lid of the bait tank open.
- Use a tank with large surface area.
- Use anti-foam. Foam decreases the surface area available for CO₂ to degas.

Salts

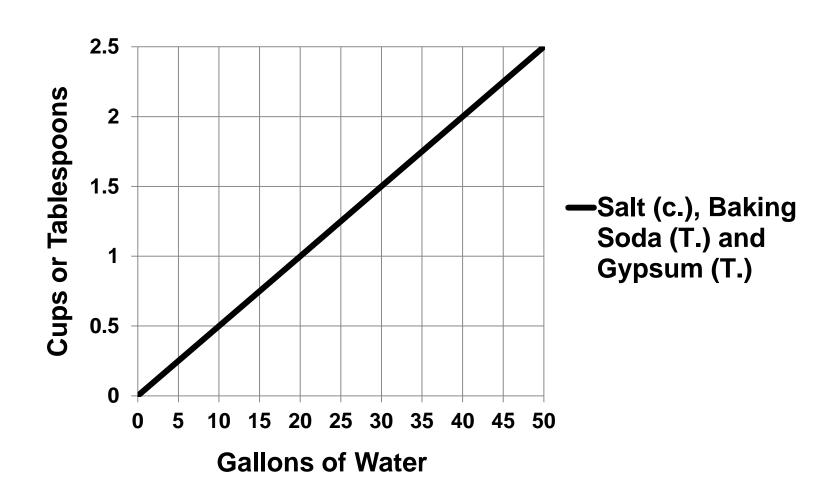
Freshwater fish expend energy to maintain salts in their bodies through a process called osmoregulation. Stress and mucus loss interferes with osmoregulation. As a result, fish lose precious salts from their bodies. Excessive salt loss can cause death by coronary arrest, blood acidification and/or ammonia accumulation. Adding salts to the bait tank water prevents your bait from losing salts.

Salts

Three different salts should be added to bait tank water:

- Sodium chloride (aka rock salt, table salt, water softener salt). DO NOT use salt that is iodized or contains additives to remove iron.
- sodium bicarbonate (aka baking soda)
- calcium sulfate (aka agricultural gypsum)

Bait Tank Chemicals



References

- G.A. Wedemeyer, Physiology of Fish in Intensive Aquaculture Systems.
- C.E. Boyd, Pond Aquaculture Water Quality Management.
- World Aquaculture, Using Salts to Transport Live Fish, 1981.
- W. Wurts, Membrane Permeability, Calcium and Osmotic Pressure, 1987.
- W. Wurts, Salts, Membrane Permeability and Biophysics, 1987.
- W. Wurts, Personal Conversation, 2013.